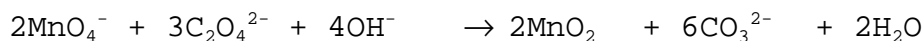
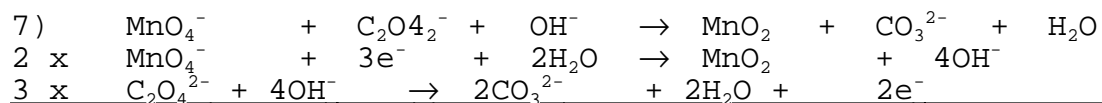
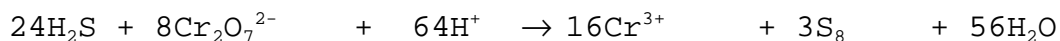
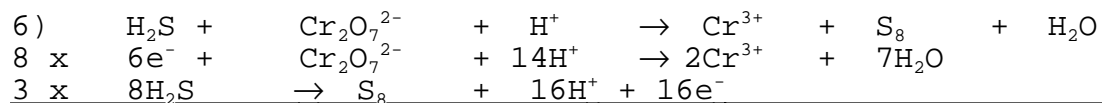
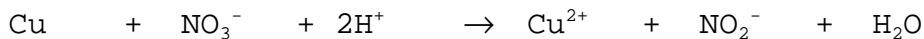
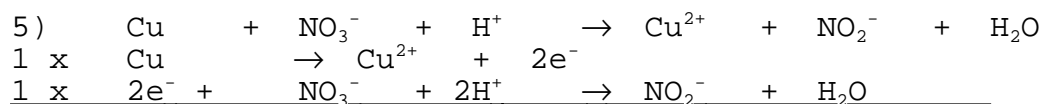
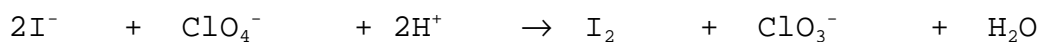
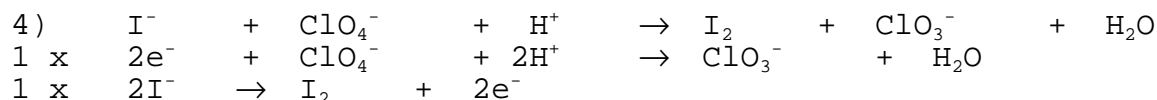
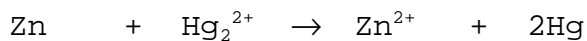
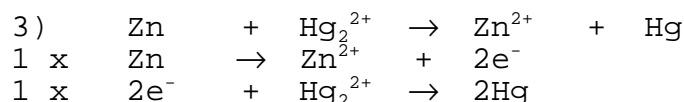
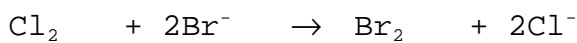
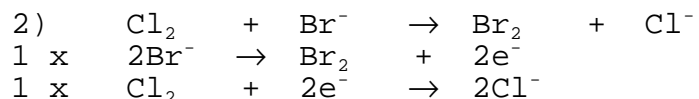
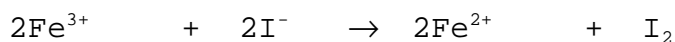
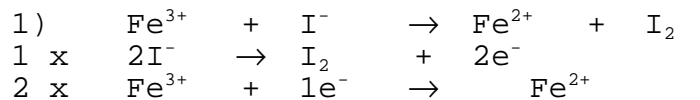
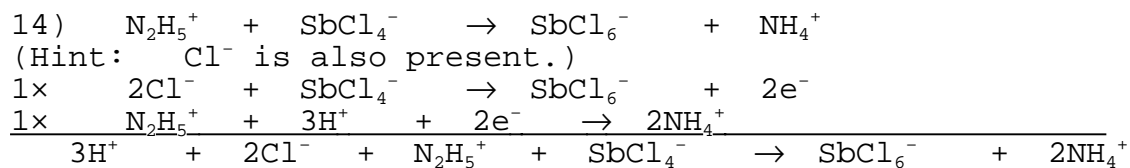
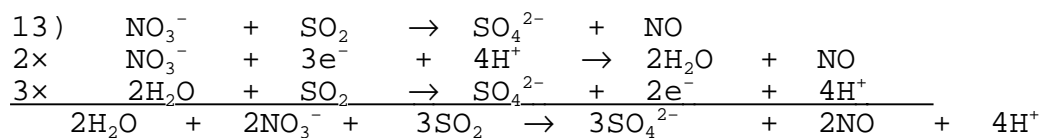
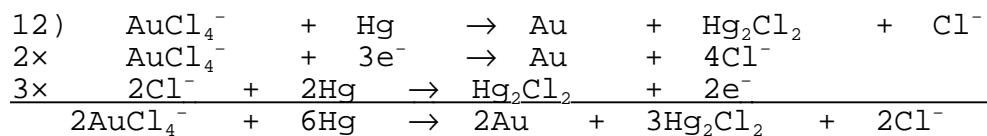
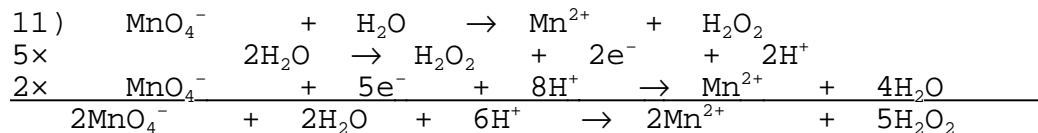
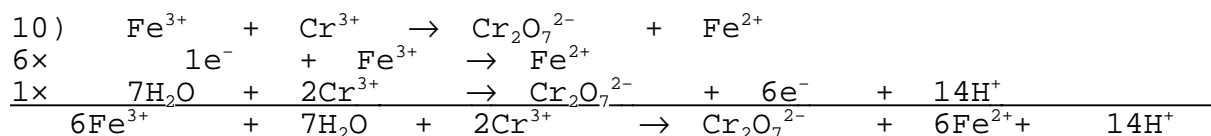
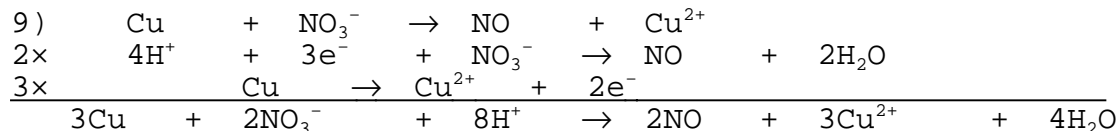
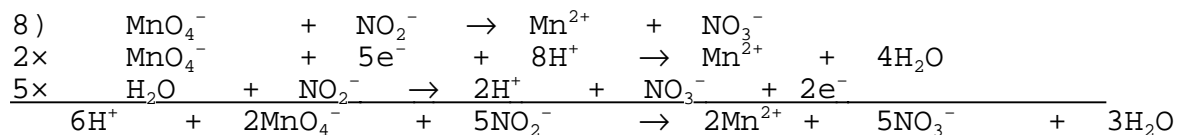


Balance the following reactions using the Half-Reaction technique. Refer to the instructions and the example of the permanganate-oxalate reaction.



Balance the following in acid solution. (To balance the oxygen, add H<sup>+</sup>s to one side of the reaction and H<sub>2</sub>O's to the other side.)



Balance the following in basic solution. (To balance the oxygen, add OH<sup>-</sup>s to one side of the reaction and H<sub>2</sub>O to the other side.)

