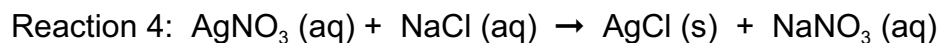
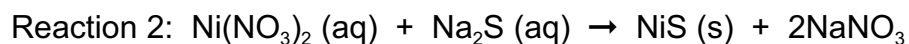


NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

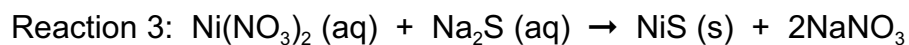
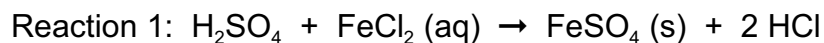
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 112

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

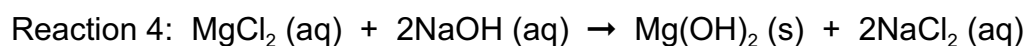
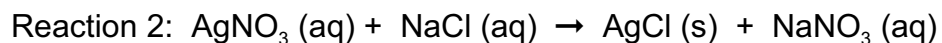
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 113

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

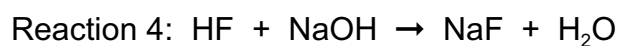
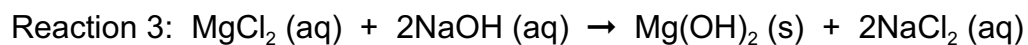
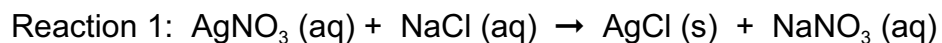
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 114

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

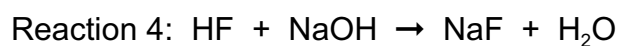
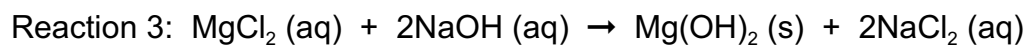
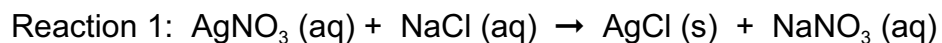
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 115

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

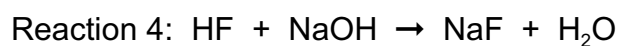
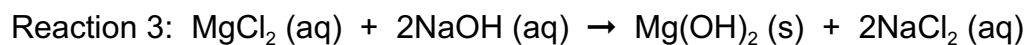
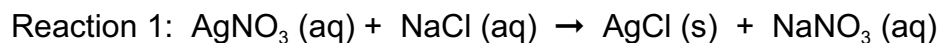
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 116

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

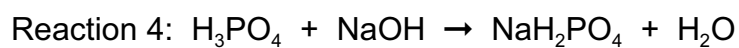
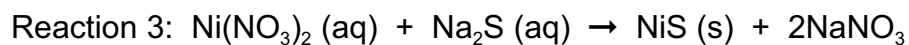
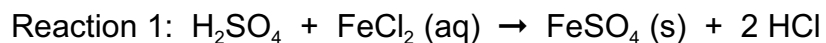
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 117

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

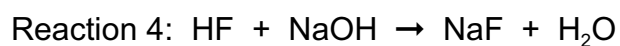
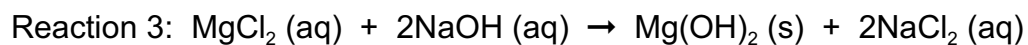
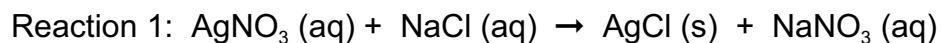
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 118

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

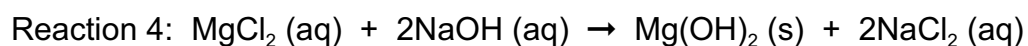
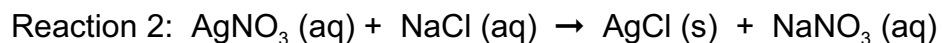
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 119

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

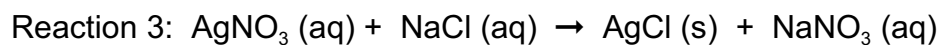
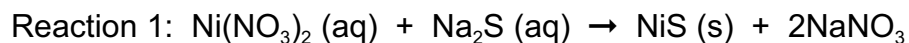
Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction
copy 120

NAME: _____

Write the net ionic reaction for each of the following reactions. ("H⁺" is an acceptable substitute for "H₃O⁺". In either case, balance the reaction.)



Reaction 1: _____ → _____

Reaction 2: _____ → _____

Reaction 3: _____ → _____

Reaction 4: _____ → _____

Homework for test 5 – Net Ionic Reaction

