

Homework for test 7 – solid-solution titration  
copy 111

NAME: \_\_\_\_\_

How many grams of  $\text{BaCl}_2$  are needed to create 24.0 mL of a 0.758 M solution?

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NAME: \_\_\_\_\_

How many grams of  $\text{BaCl}_2$  are needed to create 86.0 mL of a 0.874 M solution?

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NAME: \_\_\_\_\_

How many grams of  $C_6H_5NH_3$  are needed to create 14.0 mL of a 0.421 M solution?

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NAME: \_\_\_\_\_

How many grams of NaCl are needed to create 25.0 mL of a 0.362 M solution?

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copy 115

NAME: \_\_\_\_\_

How many grams of  $\text{BaCl}_2$  are needed to create 46.0 mL of a 0.576 M solution?

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NAME: \_\_\_\_\_

How many grams of  $\text{FeCl}_2$  are needed to create 43.0 mL of a 0.555 M solution?

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NAME: \_\_\_\_\_

How many grams of NaCl are needed to create 65.0 mL of a 0.810 M solution?

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NAME: \_\_\_\_\_

How many grams of  $\text{BaCl}_2$  are needed to create 83.0 mL of a 0.705 M solution?

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NAME: \_\_\_\_\_

How many grams of  $C_6H_5NH_3$  are needed to create 51.0 mL of a 0.274 M solution?

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copy 120

NAME: \_\_\_\_\_

How many grams of  $\text{BaCl}_2$  are needed to create 47.0 mL of a 0.348 M solution?

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copy 111	3.7876
copy 112	15.6491
copy 113	0.5548
copy 114	0.5289
copy 115	5.5165
copy 116	3.0249
copy 117	3.0769
copy 118	12.1828
copy 119	1.3154
copy 120	3.4053