

NAME: _____

How many liters of H_2S at STP are required to create 601 mL of a 0.714 M solution?

ANS: _____

NAME: _____

How many liters of HF at STP are required to create 871 mL of a 0.219 M solution?

ANS: _____

NAME: _____

How many liters of NH_3 at STP are required to create 289 mL of a 0.680 M solution?

ANS: _____

NAME: _____

How many liters of NH_3 at STP are required to create 835 mL of a 0.702 M solution?

ANS: _____

NAME: _____

How many liters of NH_3 at STP are required to create 734 mL of a 0.805 M solution?

ANS: _____

NAME: _____

How many liters of HCl at STP are required to create 802 mL of a 0.203 M solution?

ANS: _____

NAME: _____

How many liters of NH_3 at STP are required to create 452 mL of a 0.222 M solution?

ANS: _____

NAME: _____

How many liters of HBr at STP are required to create 715 mL of a 0.546 M solution?

ANS: _____

NAME: _____

How many liters of H_2S at STP are required to create 467 mL of a 0.684 M solution?

ANS: _____

NAME: _____

How many liters of CH_3NH_2 at STP are required to create 288 mL of a 0.188 M solution?

ANS: _____

Homework for test 8 – gas to concentration

copy 111 9.6122 L

copy 112 4.2728 L

copy 113 4.4020 L

copy 114 13.1302 L

copy 115 13.2355 L

copy 116 3.6469 L

copy 117 2.2477 L

copy 118 8.7447 L

copy 119 7.1552 L

copy 120 1.2128 L