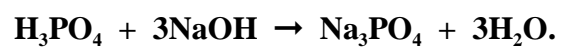


NAME: _____

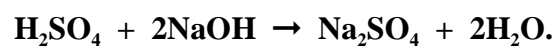
In the following reaction 37.06 mL of 0.6958 M H_3PO_4 is reacted with 27.22 mL of NaOH. What is the concentration of the NaOH?



ANS: _____

NAME: _____

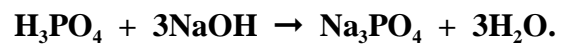
In the following reaction 38.49 mL of 0.2406 M H_2SO_4 is reacted with 15.65 mL of NaOH. What is the concentration of the NaOH?



ANS: _____

NAME: _____

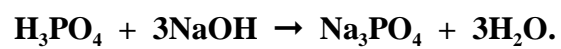
In the following reaction 36.58 mL of 0.8515 M H_3PO_4 is reacted with 30.97 mL of NaOH. What is the concentration of the NaOH?



ANS: _____

NAME: _____

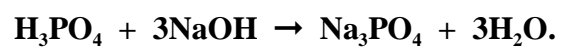
In the following reaction 22.82 mL of 0.1613 M H_3PO_4 is reacted with 40.98 mL of NaOH. What is the concentration of the NaOH?



ANS: _____

NAME: _____

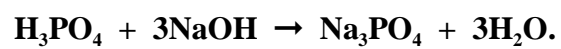
In the following reaction 49.57 mL of 0.6329 M NaOH is reacted with 34.71 mL of H₃PO₄. What is the concentration of the H₃PO₄?



ANS: _____

NAME: _____

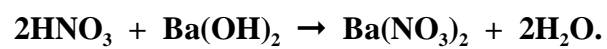
In the following reaction 47.65 mL of 0.2036 M NaOH is reacted with 36.76 mL of H₃PO₄. What is the concentration of the H₃PO₄?



ANS: _____

NAME: _____

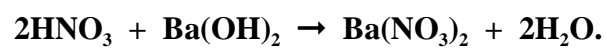
In the following reaction 441.01 mL of 0.0299 M Ba(OH)₂ is reacted with 38.35 mL of HNO₃. What is the concentration of the HNO₃?



ANS: _____

NAME: _____

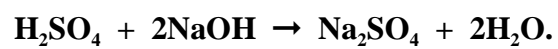
In the following reaction 415.20 mL of 0.0593 M Ba(OH)₂ is reacted with 35.93 mL of HNO₃. What is the concentration of the HNO₃?



ANS: _____

NAME: _____

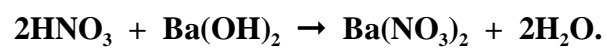
In the following reaction 26.15 mL of 0.1618 M H_2SO_4 is reacted with 12.28 mL of NaOH. What is the concentration of the NaOH?



ANS: _____

NAME: _____

In the following reaction 186.01 mL of 0.0323 M Ba(OH)₂ is reacted with 25.43 mL of HNO₃. What is the concentration of the HNO₃?



ANS: _____

Homework for test 1 – titrationcopy 200 2.8420 M

copy 201 1.1835 M

copy 202 3.0172 M

copy 203 0.2695 M

copy 204 0.3013 M

copy 205 0.0756 M

copy 206 0.6879 M

copy 207 1.3710 M

copy 208 0.6891 M

copy 209 0.4719 M