

Homework for test 6 – solution-solution titration
copy 111

NAME: _____

What volume of 3.02 M HCl solution is needed to make 403.0 mL of a 0.285 M solution?

ANS: _____

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NAME: _____

What volume of 3.66 M NaOH solution is needed to make 245.0 mL of a 0.316 M solution?

ANS: _____

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NAME: _____

What volume of 7.64 M FeCl_2 solution is needed to make 323.0 mL of a 0.593 M solution?

ANS: _____

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What volume of 6.04 M BaCl_2 solution is needed to make 541.0 mL of a 0.371 M solution?

ANS: _____

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What volume of 11.34 M NaCl solution is needed to make 167.0 mL of a 0.787 M solution?

ANS: _____

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NAME: _____

What volume of 5.92 M NaCl solution is needed to make 110.0 mL of a 0.475 M solution?

ANS: _____

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NAME: _____

What volume of 9.02 M FeCl_2 solution is needed to make 587.0 mL of a 0.752 M solution?

ANS: _____

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NAME: _____

What volume of 14.90 M NH_3 solution is needed to make 538.0 mL of a 0.848 M solution?

ANS: _____

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NAME: _____

What volume of 5.73 M HCl solution is needed to make 254.0 mL of a 0.399 M solution?

ANS: _____

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NAME: _____

What volume of 4.17 M NaOH solution is needed to make 316.0 mL of a 0.291 M solution?

ANS: _____

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copy 111 38.0 mL

copy 112 21.2 mL

copy 113 25.1 mL

copy 114 33.2 mL

copy 115 11.6 mL

copy 116 8.8 mL

copy 117 48.9 mL

copy 118 30.6 mL

copy 119 38.0 mL

copy 120 38.0 mL