

NAME: _____

What volume does 4.45 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 6.43 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 8.25 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 7.55 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 6.77 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 7.80 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 1.66 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 1.78 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 5.56 mol of H₂ gas occupy at STP?

ANS: _____

NAME: _____

What volume does 6.65 mol of H₂ gas occupy at STP?

ANS: _____

Homework for test 9 – STP

copy 111	99.6800 L
copy 112	144.0320 L
copy 113	184.8000 L
copy 114	169.1200 L
copy 115	151.6480 L
copy 116	174.7200 L
copy 117	37.1840 L
copy 118	39.8720 L
copy 119	124.5440 L
copy 120	148.9600 L