

MSC 1012 questions for Chapter 12 set 1

Molar Mass:

NAME _____

Important note: The text book uses the following terms:

atomic mass

molecular mass

formula mass

These terms are now included according to IUPAC and SIO in the terminology “molar mass”.

Give the molar mass for the following element to the nearest 0.1 g mol^{-1} .

Cl _____

As _____

H _____

Pr _____

S _____

Ta _____

U _____

Na _____

F _____

Au _____